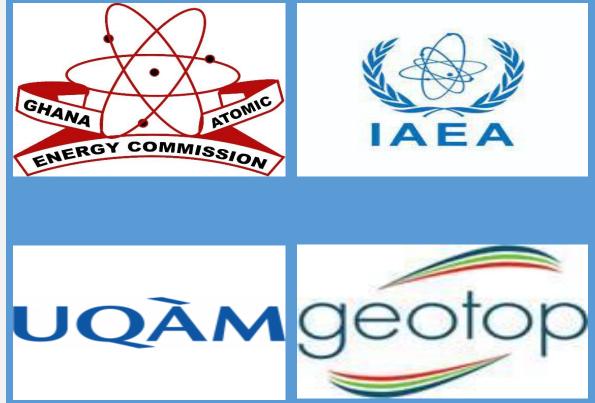
THE USE OF LEACHING EXPERIMENTS TO EVALUATE ROCK-WATER INTERACTIONS WHICH IN GROUNDWATER CHEMISTRY IN THE NORTH-WESTERN PART OF THE VOLTA RIVER BASIN OF GHANA; PRELIMINARY RESULTS.



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INTRODUCTION

chemical composition of The groundwater is derived from multiple sources, especially rock - water interactions. These activities along paths determine the final tlow composition of groundwater, for which its quality is important for end users. It's especially the case in rural areas in the North-Western part of the Volta River Basin of Ghana, where the indigenes can hardly afford treated tap water.

Although studies have been carried out to study the geological processes groundwater responsible for chemistry, none have attempted studying the results of rock – water interactions controlled under conditions (leaching experiments) which provides an elaborate means of understanding the geological responsible processes for groundwater chemistry.

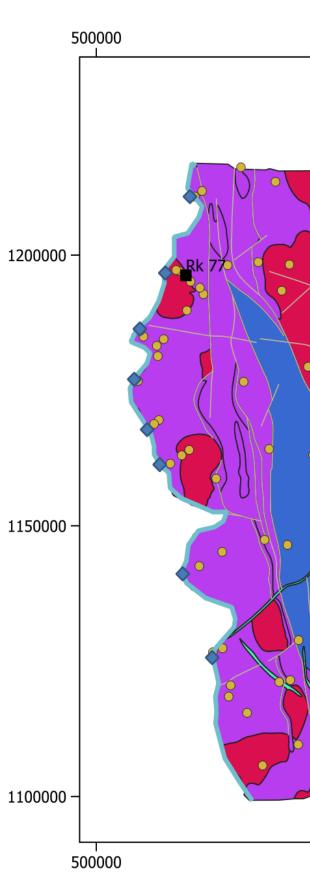
objective perform The IS to geochemical modelling on results of the leaching experiments to explain geological processes responsible for groundwater chemistry in the study area.

REFERENCES:

Bucher, K., & Stober, I. (2002). Water-rock reaction experiments with Black Forest gneiss and granite. 61–95. https://doi.org/10.1007/978-94-010-0438-1_3

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The study area Africa. In term single rainy sea Geologically, t **Crystalline Roc**





Rocks were lab after its proxin

- Thin section sample
- Samples were then crushed and a portion used for XRF and XRD analyses
- > A portion of each crushed rocks was then used for the experiments with rock: water ratio of **1:10** > Five experimental conditions were considered; Room
- H₂SO₄

STUDY AREA

ns of Climate	•••	egion of Ghana- semi-arid with short od of dry.	,	
the area is u	nderlain by F	Precambrian		100
cks (Igneous	s and metam	orphic rocks).		80
550000	600000	 Surface water samples Black Volta River 		60
•		 Black volta kivel Rock Samples Groundwater sample Faults 		% 40
Rk 90	- 1200000	Geology Birimian Supergroup Eburnean Plutonic Suite Mesozoic		20
Rk 58		'Tamnean' Plutonic Suite Tarkwaian Group		0
	k 91 - 1150000	Mauritania Mali Niger		■ Quar
Rk 38		Burkina Faso Benin Guinea Cote d'Ivory Ghana		Fig. 2.
		Liberia		Stiff dia various
550000	600000	0 10 20 k	ſM	Tempe
nap of study are	ea			
ATERIALS	S AND ME	ETHODS		
		cks representative		RK 90
of aquifers hosting groundwater				
	the area we	ere sampled.		RK 58
belled RK 90, RK 38, RK 91, RK 77 and RK 58				(RK 77)
mity to sampled groundwater				
ons were first developed from each rock				(RK 91)
				4

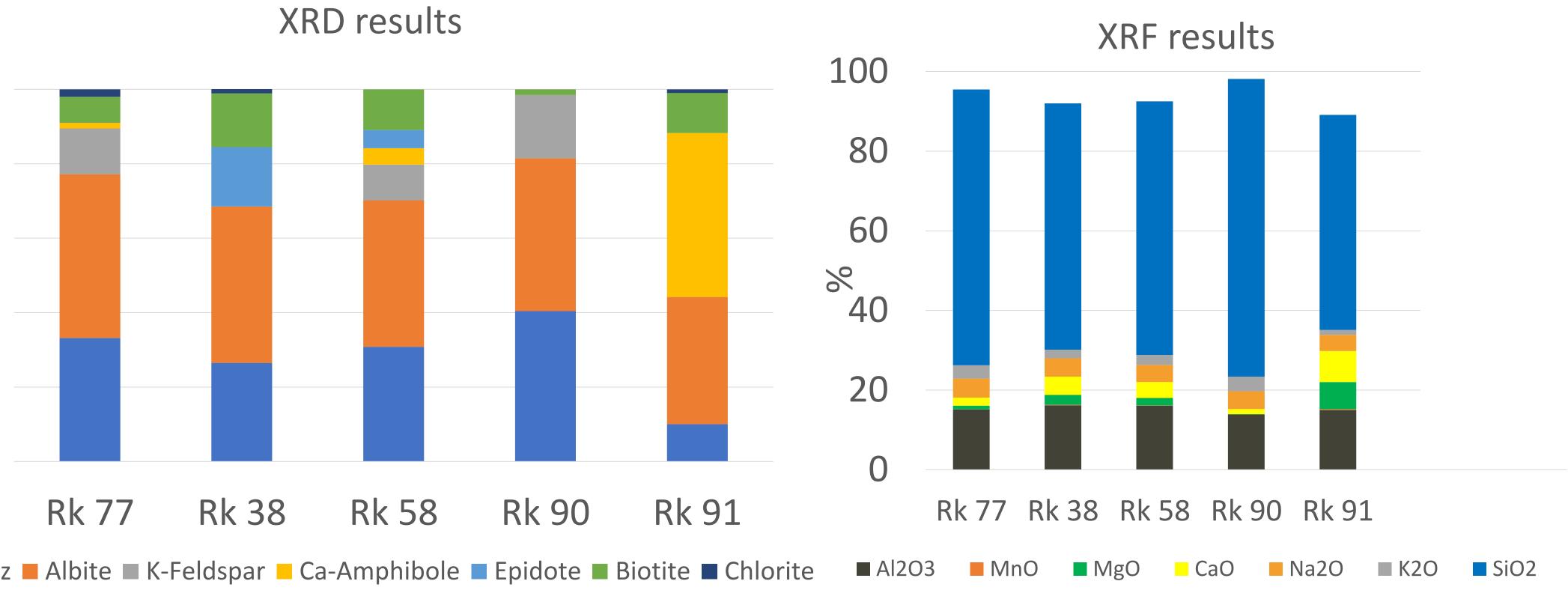
temperature, 50°C, CO₂ and a mixture of HNO₃ and

> To study the reaction progress, Electrical Conductivity (EC) and pH were measured at regular intervals while aliquot of the leachates were sampled every two months for major ion and silica analyses

Fig 3. Stiff diagrams representing the evolution of major ion chemistry of leachates from the various rock types under different experimental conditions.

RK 38

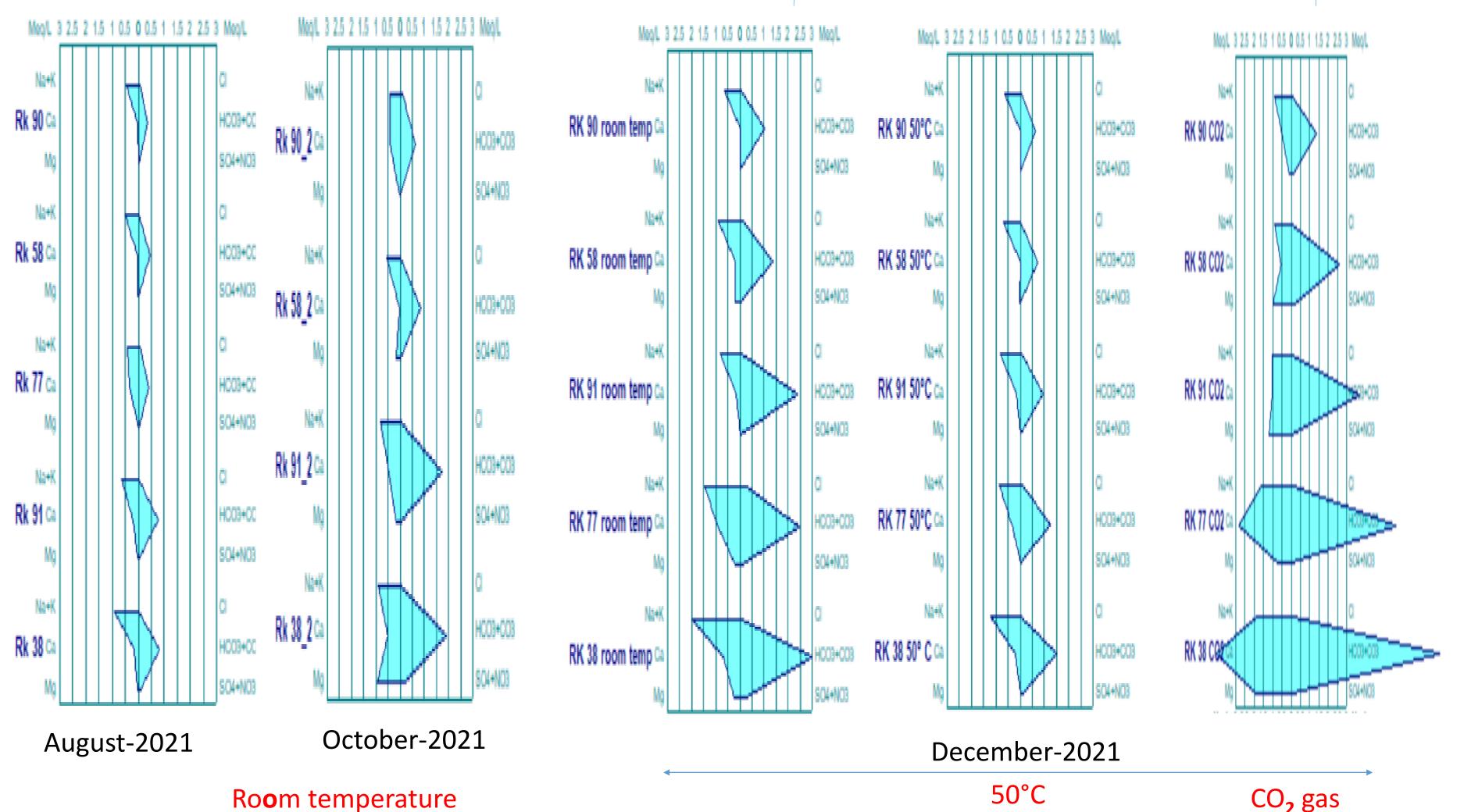
IDENTIFICATION OF MINERAL PHASES



rtz 📕 Albite 🔳 K-Feldspar 📕 Ca-Amphibole 🔳 Epidote 🔳 Biotite 🔳 Chlorite XRD and XRF results of Rock Samples used for the leaching experiment

REACTION PROGRESS OF LEACHING EXPERIMENT

liagrams of major ions plotted from results of all experiments, illustrate that leachates of each rock type produced is concentrations of water types depending on the mineral composition of rocks and experimental condition. erature did not have much influence on leachates because high temperatures reduces CO₂ At the end of the experiment During the experiment



Room temperature

CONCLUSION: Minerals phases identified under thin sections and XRD suggest that the rock types are granodioritic in composition. Higher percentages of SiO₂ and Al₂O₃ reported by XRF indicate that rocks contain more aluminosilicate minerals. Water types

Preliminary results from leaching experiments corroborates that the dissolution of minerals in groundwater is dependent on the minerals contained in the rocks, reaction time and the presence of CO_2

